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Chemistry

Buffer solutions are a mixture of
substances that have a fairly
constant pH regardless of
addition of acid or base. They are
used in medicine, industry and

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manufacturing to keep the pH of a solution stable regard- less of other reactions that might be occurring.

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20. One buffer system in the
bloodstream of animals is the
carbonic acid/bicarbonate buffer.
$$\text{H}_2\text{CO}_3 + \text{H}_2\text{O} \leftrightarrow \text{HCO}_3^- + \text{H}_3\text{O}^+$$

The carbonic acid in the
bloodstream is formed by the

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combination of carbon ...

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20. One buffer system in the
bloodstream of animals is ...

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Chemistry

Buffer solutions are a mixture of
substances that have a fairly

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constant pH regardless of addition of acid or base. They are used in medicine, industry and manufacturing to keep the pH of a solution stable regardless of other reactions that might be occurring.

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Activities for AP* Chemistry-modified 3 d) Now suppose one

keeps adding 1.0 M HCl to this buffer until a total of 101 mL of 1.0M HCl have been added. At this point, the pH of the solution will be very acidic.

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Answers Buffers

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